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I-1 CHEM 2115 Experiment # 1 INTRODUCTION TO THE LABORATORY OBJECTIVES: The goals of this session include (1) an introduction to the general chemistry laboratory and CHEM 2115 course policies and procedures, (2) safety instruction, (3) checking into lab lockers, (4) learning how to use

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### CHEM 2115 Experiment # 1 - Stockton University

1 CHEM 2115 LAB REPORT Experiment #1 Mass, Volume and Density Chem I Lab Name Jordanna Mastroddi Section # Station # A4 Date 01/24/17 1. Mass of object next to the balance: Observed mass 3.414g Posted mass 3.415g 2. Mass of 10 mL graduated cylinder: 3. Volume in graduated cylinders: 4.

### chem lab report exp 1 - 1CHEM 2115 LAB REPORT Experiment#1 ...

1 CHEM 2115 Lab Report Experiment #9 Ideal Gas Law Chem I Lab on # 006 Station # A2 Date 11/29/2016 I. Data Trial 1 Trial 2 Trial 3 Mass of Mg used (g) 0.042 0.044 0.043 Volume of H<sub>2</sub> gas collected (L) 0.04260 0.04302 0.04453 Temperature of gas (Kelvin) 294.1 294.5 294.8 Atmospheric pressure (inHg) 29.89 29.89 29.89 Height difference (mmH<sub>2</sub>O) 174.5 175.1 170.9 Vapor pressure of H<sub>2</sub>O (mmHg) ...

### 9-Gas-Law-Report.docx - 1CHEM 2115 Lab Report Experiment#9 ...

CHEM 2115 Lab Report Experiment #9 Ideal Gas Law Name Chem I Lab Section # Station # Date I. Data Trial 1 Mass of Mg used (g) Trial 3 0.0445g Trial 2 0.0420g 0.04162 L Volume of H<sub>2</sub> gas collected (L) 0.043 0.04689 L 0.046aLL Temperature of gas (Kelvin) a96K 296 K 296 K Atmospheric pressure (inHg) Height difference (mmH<sub>2</sub>O) Na mm 30 137 mmH<sub>2</sub>O Vapor pressure of H<sub>2</sub>O (mmHg) 125 mmila mmHg 21.1 ...

### Solved: CHEM 2115 Lab Report Experiment #9 Ideal Gas Law N ...

SALT Chem 321 Lecture 7 - Gravimetric Analysis Gravimetric\*Determination\*of\*Chloride\* Identifying an unknown chloride salt by gravimetric analysis GRAVIMETRIC ANALYSIS PROBLEMS - EXERCISES IN ... Chapter 12: Gravimetric Methods of Analysis Gravimetric Analysis Lab Report - Kodi Tips CHEM 2115 Experiment Two Gravimetric Determination of ...

### Gravimetric Analysis Of A Chloride Salt Lab Answers ...

Question: Due 10/ai CHEM 2115 Lab Report Experiment #3 Periodic Properties (Equations) Chem I Lab Name Section # Date Station # Using The Observations Recorded During Experiment 3 Write Balanced Equations For Any Reactions That Occurred In The Following Systems. The Equations Should Be Consistent With Your Observations. 1. Reactivity Of Metals A. I. Lithium + ...

### **Solved: Due 10/ai CHEM 2115 Lab Report Experiment #3 Perio ...**

CHEM 2115 Experiment Seven Extraction of a Copper Chelate OBJECTIVE The objectives of this experiment are: (1) to measure the concentration of copper complex in both aqueous and nonaqueous samples using a spectrophotometer. (2) to evaluate the efficiency of the extraction of copper from aqueous solutions into

### **CHEM 2115 Experiment Seven Extraction of a Copper Chelate ...**

CHEM 2115 Experiment 1 Introduction to the Analytical Balance OBJECTIVES: To acquire skills needed for the quick and accurate determination of the mass of a sample using an analytical balance. INTRODUCTION: The ability to determine mass accurately is fundamental not only to analytical chemistry but also to all of modern science.

### **CHEM 2115 Experiment 1 Introduction to the Analytical Balance**

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### **Biology Investigatory Projects For Class 12 Lastikore**

Today's experiment is an example of a single replacement reaction. It involves the measurement of the volume of hydrogen gas generated from a reaction of zinc with excess hydrochloric acid.  $Zn (s) + 2HCl (aq) \rightarrow ZnCl_2(aq) + H_2(g)$  Equation 1

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