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133 ChemQuest 44 Name: _____ Page 135 and 136: 135 6. Find the half-life for this . Page 137 and 138: 137 2. Consider the following react. Page 139 and 140: 139 Information: Calculating Equili. Page 141 and 142: 141 ChemQuest 46 Name: _____ Page 143 and 144: 143 9. Consider your answers to que. Page 145 and 146: 145 Critical Thinking Questions 5.

82 ChemQuest 28 Name:

the same exponents. Your answer will them have the same exponent as the numbers you add or. subtract. Here are some examples. Make sure you understand each step: $4.2 \times 10^6 + 3.1 \times 10^5$ make exponents the same, either a 5 or 6 $42 \times 10^5 + 3.1 \times 10^5 = 45.1 \times 10^5 = 4.51 \times 10^6$

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Answer the questions that follow based

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Answer Key

on this chemical equation: $\text{Na} + \text{MgCl}_2 \rightarrow \text{NaCl} + \text{Mg}$. Why can't NaMg be produced? (Hint: Think about the charges) Why can't NaCl_2 be produced? Are NaCl and Mg the only products that can be produced? ... ChemQuest 28 ...

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ChemQuest 35 Intro, to chemical Reoctioil\$ Date: Hour: Formula Review
1. Write the formulas for the following compounds: 2. Write the names for the following ...

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